DOCUMENT RESUME

ED 470 047 SE 066 980

AUTHOR Constabile, Kerry, Comp.; Craig, Heidi, Comp.; O'Laughlin,

Laura, Comp.; Reiss, Anne Bei, Comp.; Spencer, Liz, Comp.

What's in Your Water? An Educator's Guide to Water Ouality. TITLE

Earth Day Network, Washington, DC. INSTITUTION

PUB DATE 2002-00-00

11p. NOTE

AVAILABLE FROM Earth Day Network, 1616 P Street NW, Suite 200, Washington,

DC 20036. Tel: 202-518-0044; e-mail: education@earthday.net;

Web site: http://www.earthday.net.

Guides - Classroom - Teacher (052) PUB TYPE

EDRS PRICE EDRS Price MF01/PC01 Plus Postage.

DESCRIPTORS Biology; Chemistry; *Hands on Science; Intermediate Grades;

Junior High Schools; *Lesson Plans; Science Activities;

Science Education; *Water Quality

ABSTRACT

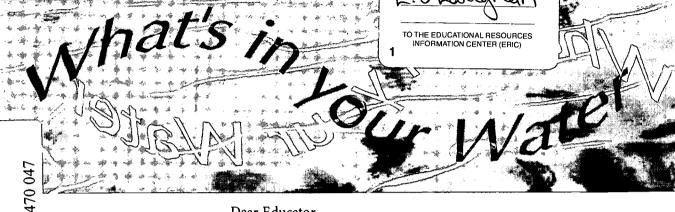
This guide provides basic information on the Clean Water Act, watersheds, and testing for water quality, and presents four science lesson plans on water quality. Activities include: (1) "Introduction to Water Quality"; (2) "Chemical Water Quality Testing"; (3) "Biological Water Quality Testing"; and (4) "What Can We Do?" (YDS)



An Educator's Guide to Water Quality







Four Lesson Plans Grades 5-8

Introduction to Water Quality

Chemical Water Quality Testing

Biological Water Quality Testing

What Can We Do?

Topics Covered:

The Clean Water Act of 1972 How to Locate Your Watershed How to Test for Water Quality Pollutants and Human Activity Point and Non-Point Pollution How to Reduce and Prevent Non-Point Pollution Community Action Opportunities

© 2002 Earth Day Network This guide may be reproduced or electronically transmitted for educational use with credit given to the publisher. Printed on recycled paper (30% post consumer waste) with soy-based ink.

Dear Educator,

This year on October 18, 2002, Earth Day Network will join with teachers, community groups, and millions of students across the country to observe the 30th anniversary of the Clean Water Act.

To help you and your students take part in this nationwide event and measure the health of your local water resources, we have produced a new teacher's guide called "What's in Your Water?" This guide provides basic information about water quality and lesson plans with hands-on activities for performing simple and fun water quality tests in lakes, streams, creeks, ponds, or other safe accessible water sites in your community.

You may either purchase an inexpensive test kit with materials for 50 tests through our website at www.earthday.net or create your own simplified version. Either way your students can report the results of their tests by entering their data on the national Clean Water Data Collection Site where they will be able to compare the health of their water with those from hundreds of sites around the nation. In addition, there are suggestions for how students can report their results to community leaders and elected local, state and federal officials.

Today, 32 years after the first Earth Day, Earth Day Network is an alliance of 5,700 organizations in 184 countries working to promote global environmental consciousness and a peaceful, just, and sustainable world. As one of the more than 95,000 educators in the United States who coordinates Earth Day events or activities, you are a vital part of the Earth Day movement. We value the work that you do and thank you for your ongoing participation and commitment to the Earth.

Please let me know what you think about this guide and your suggestions for our new program on sustainability. We invite you to check our website periodically to learn about our global campaigns, locate resources, and measure your Ecological Footprint.

U.S. DEPARTMENT OF EDUCATION Office of Educational Research and Improvement EDUCATIONAL RESOURCES INFORMATION CENTER (ERIC)

This document has been reproduced as received from the person or organization originating it.

Minor changes have been made to improve reproduction quality.

Sincerely,

Kathlein Rogers Kathleen Rogers President, Earth Day Network

Points of view or opinions stated in this document do not necessarily represent



The Clean Water Act

More than 30 years ago, the United States faced a water quality crisis. Pollutants from human, animal, and industrial sources caused Ohio's Cuyahoga River to burst into flames; oil spills on both coasts posed threats to marine wildlife and human health. Lake Erie was declared dead. Both the Potomac River in Washington, D.C. and the Boston Harbor seemed more like cesspools than waterways. Public outrage at such environmental disasters brought clean water to the forefront of the civic agenda. Congress responded in 1972 by signing the Clean Water Act, formally known as the Federal Water Pollution Control Act.

The Clean Water Act seeks to "restore and maintain the chemical, physical, and biological integrity of the

nation's water." Consequently, states adopted uniform minimum water quality standards, monitored by the then newly-created Environmental Protection Agency (EPA). Today, reduced toxic flows have lowered fish kills allowing lakes and rivers to revive. Municipal wastewater treatment plants, supported by billions of dollars in federal investments serve almost 190 million people, 50 million more than in the late 1960s.

The Clean Water Act guarantees us clean, safe water for drinking, fishing, and swimming. Nevertheless, it is important, to understand that 40% of our water sources still do not meet healthy standards.



Source: United States Public Interest Research Group, Clean Water Network

Watersheds

How to locate your watershed:

The EPA provides a useful site, where students can enter their home information, from zip code to tribal nation, locate and learn about their local watershed at:

http://cfpub.epa.gov/surf /locate/index.cfm

A watershed is the region that draws water and snowmelt into bodies of water. Every waterway, whether a small tributary, stream, or lake, has its own associated watershed.

Watersheds can be large or small and can extend across county, state, and national boundaries.

In addition to acting as drainage basins, watersheds capture precipitation, filter, and store water. Within each watershed there is a diverse system of ecoregions, marine life, and air sheds. Everyone living and working within a watershed needs to cooperate to ensure healthy conditions because water moves downstream in a watershed. Any activity that negatively affects water quality will change the characteristics of the water downstream as well, and impact on the water quality of the body of water.

Simple choices in daily activities that can affect watershed health include: dumping used motor oil down a sewer; over-fertilizing lawns and gardens;

applying sand and chemicals to driveways and sidewalks; removing vegetation such as plants, trees, and grasses along a riverbank; leaving pet waste on the ground; and poorly maintaining home septic systems. Communities as a whole can also affect the water quality of a watershed through land use decisions such as where to locate housing, shops, factories, parks, and farms. By analyzing a watershed's physical charact4ristics and land use patterns, students will understand how their watershed is changing and what they can do to protect it. Raising awareness by involving individuals and their communities creates a collaborative effort to protect home watershed areas. Students can play a powerful role in this process.

ERIC*

Source: Adapted with permission from Earth Force. For more information, see www.green.org

Testing for Water Quality Field Experience (with or without kit) *

Water quality testing determines whether water is safe for different types of use such as swimming, fishing, drinking, and irrigation. Knowledge of the water quality within your watershed provides understanding about human activities and our role in the ecological processes. Typically, tests for water quality identify several indicators that can be used to determine the health of a watershed. Key indicators include alkalinity, dissolved oxygen, nitrates, pH, temperature, and turbidity. A glossary of key terms follows:

Alkalinity measures the ability of the water to neutralize (or buffer) acids and keep the pH from changing. **Sources:** Rocks, soils, salts, plant activities, and certain wastewater discharges. **Effects and Hazards:** High water alkalinity causes higher algae and plant growth; while low alkalinity indicates that the water's ability to buffer acids is poor. If there are drastic changes in alkalinity, many chemical and biological processes will be affected.

Dissolved oxygen measures the presence of oxygen gas molecules in water. These oxygen molecules keep organisms living, sustain species reproduction, and support many chemical processes that occur in water. Water that maintains high dissolved oxygen levels is generally considered environmentally healthy; although saltwater, warm water, and water at high altitudes can contain less dissolved oxygen and still be part of a health-sustaining ecosystem. **Effects and Hazards:** Low dissolved oxygen levels stress fish and other aquatic organisms.

Nitrates are essential for plant growth, although too much nitrate may indicate a pollution problem. **Sources:** Soil, animal wastes, and decomposing plants; sewage, fertilizers, and animal waste. **Effects and Hazards:** High levels of nitrates affect dissolved oxygen levels and lead to excessive plant growth, affecting the types of plants and animals that can live in the water. Infant blood poisoning, cancer, and genetic changes have been attributed to high levels of nitrates.

pH measures the acidity of a solution as an "index" of the amount of hydrogen ions present in a substance and affects many chemical and biological processes. **Sources:** Acidity increases due to mine draining, industrial waste, and acid precipitation.

- pH is measured on a scale of 0-14, with a neutral pH at 7
- A pH less than 7 is an acid, with more hydrogen ions
- A pH greater than 7 is basic, and has more hydroxide ions
- Most natural water has a pH value between 5.0 and 8.5. Rainwater has a pH between 5.5 and 6.0. Salt water has a pH between 8.0 and 8.5
- Most aquatic animals prefer a range of 6.5 to 8.0
- All water with a pH of less than 5.0 or greater than 8.5 should be viewed as suspicious

Temperature measures the degree of heat in the water which affects the rate of many of the waterways' biological and chemical processes and the amount of dissolved oxygen. **Sources:** Air temperature, the amount of runoff, the temperature of water running into the waterway, amount of sunlight, and water cloudiness. **Effects and Hazards:** Temperature affects the rate of photosynthesis and decomposition in plants. High temperatures may be a sign of thermal pollution from industrial sites.

Turbidity is the clarity of the water. Clear water has a low turbidity while murky water has a high turbidity. **Sources:** Small particles suspended in water such as algae, clay, microorganisms, silt, organic chemicals, decaying vegetation, or chemical wastes. **Effects and Hazards:** Turbidity can interfere with the process of disinfecting water. Particles may absorb or bond with toxic substances and prevent their removal during treatment.

Adapted from:

http://www.beesinc.org/resource/currenha/watmonit.htm

http://wilkes.edu/~eqc/helpguide.htm

* Note: If you have purchased the NWMD kit, you can follow the instructions included with it.



Pollutants and Human Activities

There are many different sources of water pollution. Water quality changes quickly as water composition is altered by various ground surfaces over and under which it flows and combines with rock, minerals, other elements and numerous materials which are a direct result of human activities. The latter will be of particular importance to students as it relates to how their daily lives can affect our water sources.

Point and Non-point Source Pollution

Pollution that comes from a single, identifiable source, such as a factory or discharge from a sewage treatment plant, is called point source pollution. Once the source is identified it becomes easier to improve water quality. Non-point source (NPS) pollution is attributable to diverse sources. The EPA reports that NPS pollution is a leading cause of water quality problems. NPS is primarily caused by rainfall or snowmelt moving over and through the ground causing runoff to pick up and carry away natural and human-made pollutants. This runoff finally deposits the pollutants into lakes, rivers, wetlands, coastal waters, and even our underground sources of drinking water.

Example Pollutants

- Excess fertilizers, herbicides, and insecticides from agriculture and residential areas
- Oil, grease, and toxic chemicals from urban runoff and energy production
- Sediment from improperly managed construction sites, crop and forest lands, and eroding stream banks
- Salt from irrigation practices and acid drainage from abandoned mines
- Bacteria and nutrients from livestock, pet wastes, and faulty septic systems

How to Reduce and Prevent Non-point Source Pollution

Some activities are federal responsibilities such as ensuring that federal lands are properly managed to reduce soil erosion. Some are state responsibilities: for example, developing legislation to govern mining and logging and to protect groundwater. Others are handled locally, through zoning and erosion control ordinances. As citizens, we play an important role by practicing conservation and by making it our personal business to properly dispose of home hazardous water, using non-polluting lawn and garden supplies, recycling plastics and other detrimental debris. Finally instilling an awareness which will engender careful and conscientious practices in our daily activities both personally and through our community programs does much to assist in controlling NPS pollution and will ultimately safeguard our watershed.

Source: Environmental Protection Agency (http://www.epa.gov/owow/nps/qa.html)

Preparation for Lesson Plans

You may purchase an inexpensive test kit with materials for 50 tests through our website at http://www.earth-day.net/ goals/clean_water.stm or by calling 1-800-344-3100 ext. 7015. You may also create your own kit. Either way, your students can report the results of their tests by entering their data on the national Clean Water Data Collection Site. Registration should be done in advance, on the web page listed above, so that you can compare your results with those from thousands of sites around the country. If possible, plan ahead to arrange a field trip or determine an on-campus outdoors site in which to perform these water quality tests. If this is not possible, give yourself and your students at least one week to bring in samples from an appropriate source. These sources include lakes, streams, ponds, creeks, or ocean—not puddles, tap water, or storm drains.

Recommended Internet Resources can be found on our website at http://www.earthday.net/goals/clean_water.stm.



esson Plan 1: Introduction to Water Quality

OVERVIEW:	KEY ISSUES/ CONCEPTS	80 S
In this collaborative, hands-on activity	Water Quality	- 4
students learn about clean water and how	Chemical Testing	· · · · · · · · · · · · · · · · · · ·
to test for it. Using historical information,	Biological Testing	4 4
scientific background, investigation and	Relative Impacts	
hands-on experimentation, students will		
learn about their role in the water cycle,	SUBJECT AREAS	
and how human impacts can both	Chemistry, Biology	3 9
positively and negatively affect water	Environmental Studies	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
squality in local, regional, national and	History, Geography	0 000
worldwide watersheds.		
	GRADE LEVEL : 5-8	3 Jan 19
	经接一	

Setting: Indoor

Class Size: 50 and under

Preparation: Register water testing site (if doing field water testing) on Year of Clean Water national water quality monitoring website: http://www.earthday.net/goals/clean_water.stm

Materials: Newspapers and magazines with articles relative to water quality, reference books, in-class Internet access if possible.

Time: 45 minutes to 1 hour

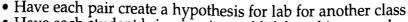
Objectives:

- Describe water, watersheds and the water cycle, human interactions and their effect on the water cycle
- Identify local watershed(s) using research skills through books, magazines and
- Understand the importance of clean water in the historical context of the Clean Water Act

Activities:

- 1. Discussion (15 minutes) Use this time to introduce topics and discussion about water
 - What does water do for us?
 - Where does it come from?
 - What would happen if it were polluted? Define polluted.
 - Why is it important to have clean water? What is clean water?
 - Does our state or community have certain standards for water quality?
- 2. Research (20-25 minutes) Put students in groups of two and assign each group a question. Bring in newspapers, magazines with pertinent articles. Have students use Internet (see Recommended Internet Reources on our website, http://www.earthday.net/goals/ clean_water.stm) encyclopedias, and other media to research water pollution. If possible, invite a local water expert to present information.
- 3. Closure (5-10 minutes) Share information. Assign homework:

BEST COPY AVAILABLE



• Have each student bring in a jar and lid for taking samples





Lesson Plan 2: Chemical Water Quality Testing

OVERVIEW:

This is a hands-on activity where students learn about key chemical indicators of water quality and about the health of their own watershed through field experience.

Students collect water samples, analyze the quality of the collected water, and test their predictions, while also learning about how humans affect the water cycle.

KEY ISSUES/CONCEPTS:

Water Quality
Chemical Testing
Scientific Method

SUBJECT AREAS:

Chemistry, Biology, Environmental Studies, Earth Science

GRADE LEVEL: 5-8

Setting: Indoor and outdoors Class Size: 50 and under

Materials: Clean Water Testing Kits or thermometers, Secchi disk, pH strips, dissolved oxygen test tabs, glass jars to collect additional samples for biological testing during next class, samples of different types of water, some airtight, some not.

Time: 45 minutes- 1 hour (allow more time if testing outdoors)

Objectives:

- Determine what makes water of good or poor quality
- Discuss and define key indicators to water quality
- Gain scientific field experience while collecting water samples
- Use test to evaluate and determine the quality of water sampled

Activities:

- **1. Discussion** (5-10 minutes) Students share results from research. Discuss testing process. Information can be found in the kit directions and lab rules.
- 2. Water testing (25 minutes) If testing outdoors, assign pairs to a specific spot to sample. Try to test in many different areas to ensure diverse results. If possible, schedule as a field trip to increase available time and have students sample several different areas and compare the results. Have students take additional samples in jars for biological water quality testing during next class period.

If testing indoors, students should have brought in samples. Make sure you have some pond water or ocean water samples they can compare with their tap water.

- 3. Closure (5 minutes) Assign homework:
 - Students document their lab findings either in narrative format or as a laboratory write-up
 - Create predictions for biological water testing. Have students research the organisms and species that live or have lived in their watershed

BEST COPY AVAILABLE



Lesson Plan 3: Biological Water Quality Testing

OVERVIEW:	KEY ISSUES/CONCEPTS:	
This is a hands-on activity where students:	* 652. ·	
learn about key biological indicators of		li 🕶 🤌 🦓
water quality and about the health of their		t e e e
own watershed through field experience.		
Students collect water samples, analyze the	0,	
quality of the collected water and test their	The state of the s	
predictions, while also learning about how		
humans affect the waer cycle.	GRADE LEVEL : 5-8	
		· · · · · · · · · · · · · · · · · · ·

Setting: Indoor lab activity Class Size: 50 and under

Materials: Hand-held magnifying glasses, lab notebooks, eyedroppers, microscopes, slides/cover slips, samples from previous testing of different types of water, dichotomous key or pictures of common organisms found in like water sources.

Time: 45 minutes- 1 hour

Objectives:

- Apply knowledge from chemical testing to create predictions of the impact of chemical water health on biological water quality
- Relate biological water health to learned historical impacts of poor water quality

Activities:

- 1. Discussion (5-10 minutes) Students share predictions and findings from research. Discuss observation process. Distribute hand-held magnifying glass, instruct on use of microscopes if necessary (i.e., lenses, cover slip fragility, etc.).
- 2. Biological water testing (25 minutes)
 - Have students examine water with untutored eye, noting presence of plants or visible aquatic organisms, marking observations in lab book
 - Examine under hand-held magnifying glass, marking observations in lab book
 - Using an eyedropper, students put a few drops of water on a slide, topping with a cover slip. Examine under microscope, noting observations in lab book

If students brought in their own samples, make sure you have some pond water or ocean water samples for making comparisons.

- 3. Closure (5 minutes) Assign homework:
 - Students document their lab findings either in narrative format or laboratory write-up
 - Students should think of ways they can improve the health of their water
 - Encourage research about what others have done to improve the quality of water in their communities



Lesson Plan 4: What Can We Do?

KEY ISSUES/CONCEPTS: OVERVIEW: Water Quality In this follow-up exercise students work together to interpret the results of water: Pollutants quality testing. They learn about the Personal Impact sources and effects of pollutants on water **SUBJECT AREAS:** health and identify how they personally affect water quality and how they can work Environmental Studies, Social Studies, Language Arts, Art to improve it **GRADE LEVEL: 5-8**

Setting: Indoor

Class Size: 50 and under

Materials: Art materials, researching materials, reports on water quality from your local water company, local environmental groups, and/or State Environmental Protection Department.

Time: 45 minutes

Objectives:

- Describe pollutants, their sources and their effects on biological and chemical health of water
- Generate an overall idea of the health of local watersheds, using results, consensusbuilding and brainstorming skill
- Identify how students affect water quality through daily interactions affecting the water cycle
- Identify opportunities for civic action on a personal, local, national and international level

Activities:

- 1. Discussion (5-10 minutes) Compile all hypotheses and results as a class and brainstorm about what the results mean. Translate into real world language.
- 2. Assignments (20 minutes) Suggested projects for service learning (in groups of 3-4):
 - Field trip to water plants and reservoirs, especially if outdoors testing was not possible
 - Write a persuasive piece about why it is important to test water (in form of letter or report)
 - Turn lab reports into art and/or presentation projects, posters, oral and/or written reports, displays on the status of your local watershed, incorporating student results and what students think needs to be done to improve local water quality
- 3. Closure (5 minutes) Continue classroom assignments as homework (essay, artwork, report).



Suggestions for Civic Action Get the Word Out! Students share their knowledge of local water quality

Not all members of your community have the opportunity to participate directly in water quality testing. In fact, they may not even be aware of the potential threats to their watershed's health. Take advantage of student knowledge to educate the community.

Possible activities include:

- Media outreach: Have students write letters to the editor regarding the health of their water shed, including simple suggestions for improvement. Or, working in groups, have students write and distribute a press release to local media in order to encourage reporters to cover water quality and your class' activities.
- Community art exhibit: Find a community sponsor (library, grocery store, shopping mall, etc.) where students can hang their awareness-building artwork in a high traffic area. Include projects that highlight not only the problems, but also possible solutions.
- Political action: Your students may not be old enough to vote, but that doesn't mean they can't have a voice. Have students meet with government officials or attend town hall meetings and raise the issue of water quality. Or take advantage of campaign season to highlight a candidate's position on watershed protection and other environmental issues.

Community Water Testing Night How clean is our water? What can we do?

This activity will not only raise awareness, but will also give community members the opportunity to participate in testing similar to what the students did. Invite parents, media (students might draft a media advisory) politicians, water experts, etc.

- Include a water quality testing station so visitors can test water samples themselves. Identify the source on a map
- Students can present reports, show posters and art projects
- Encourage citizens to take action similar to the suggestions under "Get the Word Out!"
- If local politicians attend your event, try to organize a follow-up forum with city or county leadership, including possible introduction of legislation to protect the local watershed

Storm Drain Stenciling

If you determine trash from storm drains to be a possible pollutant in your community, this is a great project for students. Students mark drains with stencils that advertise that all dumped materials flow into the local watershed. The storm drains provide a long-term reminder that everyone is responsible for protecting their watershed's health. Post flyers in the stenciled neighborhoods, allowing students to share their findings with the community. This also provides a long-term reminder for everyone. The flyer may also be a good place to advertise your Community Water Testing Night.



Earth Day Network 1616 P Street NW, Suite 200 Washington, DC 20036



What's in Your Water?

Introduction to Water Quality

Chemical Water Quality Testing

Biological Water Quality Testing

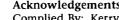
What Can We Do?

Join us in Celebrating the Year

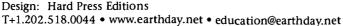
News from Earthday Network

We are developing a year-round educational program, focused on sustainability with opportunities for teaching the skills of civic action and environmental citizenship. This guide marks the first step in this exciting new direction.

BEST COPY AVAILABLE



Complied By: Kerry Constabile, Heidi Craig, Laura O'Laughlin, Anne Bei Reiss, Liz Spencer. Special Thanks: Laura Bullock, Erin Constabile, Sarah Faulkner Leff. Illustration: Ayla Gams









DOCUMENT IDENTIFICATION:

U.S. Department of Education

Office of Educational Research and Improvement (OERI)
National Library of Education (NLE)
Educational Resources Information Center (ERIC)



REPRODUCTION RELEASE

(Specific Document)

	idi (Raig, Laura O'Laughlin, F	Thire bei Reiss, he sperieur	
edrograte Source:		Publication Date:	
Earth Day N	letwork_	9/2002	
" DEPROPULATION RE	LEACE.	·	
II. REPRODUCTION RE			
announced in the monthly abstract journal of	possible timely and significant materials of interest the ERIC system, Resources in Education (RIE), are	e usually made available to users in microfich	
reproduced paper copy, and electronic media of each document, and, if reproduction rele	a, and sold through the ERIC Document Reproduction ase is granted, one of the following notices is affixe	n Service (EDRS). Credit is given to the sourced to the document.	
	d disseminate the identified document, please CHEC		
at the bottom of the page.	u disserninate the identified document, pieces on Ex	on one of the second and order	
The semple sticker shown below will be effixed to all Level 1 documents	The semple sticker shown below will be affixed to all Level 2A documents	The sample sticker shown below will be affixed to ell Level 2B documents	
PERMISSION TO REPRODUCE AND DISSEMINATE THIS MATERIAL HAS BEEN GRANTED BY	PERMISSION TO REPRODUCE AND DISSEMINATE THIS MATERIAL IN MICROFICHE, AND IN ELECTRONIC MEDIA FOR ERIC COLLECTION SUBSCRIBERS ONLY, HAS BEEN GRANTED BY	PERMISSION TO REPRODUCE AND DISSEMINATE THIS MATERIAL IN MICROFICHE ONLY HAS BEEN GRANTED BY	
sample	Sample	Sample	
TO THE EDUCATIONAL RESOURCES INFORMATION CENTER (ERIC)	TO THE EDUCATIONAL RESOURCES INFORMATION CENTER (ERIC)	TO THE EDUCATIONAL RESOURCES INFORMATION CENTER (ERIC)	
1	2A	2B	
Level 1	Level 2A	Level 2B	
Check here for Level 1 release, permitting reproduction end dissemination in microfiche or other ERIC erchival medie (e.g., electronic) and paper copy.	Check here for Level 2A release, permitting reproduction and dissemination in microfiche end in electronic medie for ERIC archivel collection subscribers only	Check here for Level 2B release, permitting reprodu and dissemination in microfiche only	
If permis:	Documents will be processed as indicated provided reproduction of sion to reproduce is granted, but no box is checked, documents will	quality permits. be processed et Level 1.	

Washington DC 20036



Earth Day Network. 1616 PSt. NW, Suit 200

III. DOCUMENT AVAILABILITY INFORMATION (FROM NON-ERIC SOURCE):

If permission to reproduce is not granted to ERIC, or, if you wish ERIC to cite the availability of the document from another source, please provide the following information regarding the availability of the document. (ERIC will not announce a document unless it is publicly available, and a dependable source can be specified. Contributors should also be aware that ERIC selection criteria are significantly more stringent for documents that cannot be made available through EDRS.)

Ear	th Day Network
Address:	, J.,
WWW.	earthday.net/goals/earth_water_guide.pdf
Price:	el
IV. RE	FERRAL OF ERIC TO COPYRIGHT/REPRODUCTION RIGHTS HOLDER:
If the right to address:	grant this reproduction release is held by someone other than the addressee, please provide the appropriate name and
Name:	
Address:	
V.	WHERE TO SEND THIS FORM:
Send this for	m to the following ERIC Clearinghouse:

However, if solicited by the ERIC Facility, or if making an unsolicited contribution to ERIC, return this form (and the document being contributed) to:

ERIC Processing and Reference Facility 4483-A Forbes Boulevard Lanham, Maryland 20706

> Telephone: 301-552-4200 Toll Free: 800-799-3742 FAX: 301-552-4700

e-mail: info@ericfac.piccard.csc.com

WWW: http://ericfacility.org

EFF-088 (Rev. 2/2001)

Publisher/Distributor:

